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Research Article





Long-Term Stability Study of Complex Darunavir: β-Cyclodextrin

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Abstract

Darunavir, a protease inhibitor most widely used in the treatment of HIV infection, was complexed with β -cyclodextrin due to the low solubility in water and its poor bioavailability. This research describes the study the long-term stability of the complex darunavir: β -cyclodextrin which was kept in a climatic chamber for 24 months at 30°C ± 2°C and 75 % UR ± 5 %. The samples were analysed using LC-MS, 250 mm × 4.6 mm CN Luna column, water + 0.1 % glacial acetic acid: acetonitrile + 0.1 % glacial acetic acid 60:40 (v/v) as mobile phase, flow rate of 1.0 mL min⁻¹, UV detection at 268 nm and ambient room temperature (25°C) before the start of study and thereafter 3, 6, 9, 12, 18 and 24 months. The data obtained associated with the infrared, TG, DSC and X-ray diffraction analysis was sufficient to study the behavior of complex darunavir: β -cyclodextrin. The results of this research indicate that the stability of the complex darunavir: β -cyclodextrin is high under conditions associated of temperature and humidity.

Keywords: β-cyclodextrin; Darunavir; Degradation products; LC-MS; Long-term stability

Introduction

Darunavir is a protease inhibitor used in the treatment of Human Immunodeficiency Virus (HIV) infection. It is a pillar of therapy cocktail for patients with the virus^[1]. Like most new drugs it presents low water solubility and poor bioavailability, leading to poor absorption in the gastrointestinal tract^[2]. These new drugs require frequent administration at relatively high doses, being the major cause of non-adherence to treatment and an obstacle for compliance of pharmacotherapy^[3]. For this reason the complexation of darunavir to β -ciclodextrina was performed.

The interactions of drugs with cyclodextrin is an important feature in the pharmaceutical field since these systems can modifying their chemical stability as well as other properties such as solubility, dissolution rate and bioavailability^[4]. β -cyclodextrins are the most used for the development of pharmaceuticals, particularly because of their complexing properties, which provide increased solubility and consequent increased dissolution rate of poorly soluble drugs^[5].

There is no long-term stability method described in the literature for complex darunavir: β -cyclodextrin. Its behavior under conditions of temperature and

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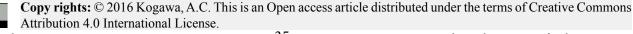
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humidity are unknown and, therefore, the development of a formulation and study of this new presentation becomes blind.

Stability is a required quality for all drugs and consists of the period from the date of manufacture and packaging of the formulation until their chemical and biological activity is not less a predetermined level of labeled content and its physical properties have not changed appreciably and deleterious fashion^[6].

The study of the stability of drugs and medicines is mandated by regulatory agencies throughout the world. The parameters used in the study of accelerated and long-term stability are indicated by the ICH^[7], the guide to conducting stability



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studies of ANVISA^[8] and by the guidance of the World Health Organization^[9].

The stability of pharmaceuticals depends on environmental factors such as temperature, humidity and light, factors related to the product itself and factors related to packaging materials^[8].

Monitoring the stability of drugs is one of the most effective methods for evaluation, prediction and prevention of problems related to the quality of the product during the validity.

The long-term stability study checks the physical, chemical, biological and microbiological characteristics of a pharmaceutical product during and, optionally, after the expiration date expected^[8].

This research describes the long-term stability study of complex darunavir: β -cyclodextrin maintained in a climatic chamber for 24 months at 30°C ± 2°C and 75 % UR ± 5 %. Analyzes were performed by LC and LC-MS at times 0, 3, 6, 9, 12, 18 and 24 months to evaluate the content of darunavir and appearance of degradation products. The time 0 and 24 months were also compared through the infrared, TG, DSC and X-ray diffraction techniques.

Experimental

Instrumentation and reagents

Stability of darunavir:β-cyclodextrin

Equipments: The samples were degraded in a climatic chamber (MarconiTM). LC analysis was performed on a WatersTM LC system equipped with Waters 1525 binary gradient chromatography pump, Rheodyne Breeze 7725i manual injector and Waters 2487 UV-Vis detector, PhenomenexTM CN Luna (250 mm × 4.6 mm, 5.0 µm particle size) column. LC-MS analysis was performed on ShimazduTM HPLC system connected to AmaZon SL BrukerTM ion trap mass spectrometer operating in positive ion electrospray ionization mode. Analytical balance model 410 KernTM, ultrasonic bath Ultrasonic Cleaner UniqueTM, water purification system MilliporeTM and membranes of polytetrafluoroethylene (PTFE) hydrophilic with pore 0.45 µm and diameter 47.0 mm were used.

In IR analysis was used spectrophotometer IR Prestige-21 ShimadzuTM. The TG curves were obtained from the SDT Q600 V8.3 Build 101 and the DSC curves with heat flow were obtained from the DSC Q100 TA InstrumentsTM. In the X-ray diffraction was used Rigaku rotating anode RINT2000 equipment, with curved graphite monochromator in the diffracted beam, divergence slit and scattering of 0.25°, receiving slit of 0.3 mm and Soller slit of 2.5° of divergence in CuK α radiation ($\lambda = 1.5406$ Å).

Chemicals and reagents: The chemical used were HPLC grade acetonitrile (J.T.BakerTM), HPLC grade glacial acetic acid (SynthTM), analytical reagent grade ethyl alcohol (SynthTM), deionised Milli Q water (MilliporeTM). The complex darunavir: β -cyclodextrin in powder^[9] developed by our research group was used. Mobile phase was prepared by mixing water + 0.1 % glacial acetic acid and acetonitrile + 0.1 % glacial acetic acid in the ratio 60:40 (v/v) filtered through 0.45 µm membrane filter. Diluent was ethyl alcohol. A stock standard solution equivalent to 1000 µg mL⁻¹ darunavir was prepared by dissolving an accurately weighed amount of pure drug in the diluents.

Methods

Long-term stability study protocol

To study long-term stability of complex darunavir: β -ciclodextrin in powder was used climatic chamber 835/UR MA (MarconiTM) with controlled temperature of 30°C ± 2°C and 75% UR ± 5% UR. In times of 3, 6, 9, 12, 18 and 24 months, samples were taken to assess the behavior of the drug to complexed β -cyclodextrin. Identify possible degradation products by LC-MS and evaluate the content of darunavir by LC. The chromatograms were compared together and with an analysis of time zero (drug was prepared and immediately analyzed) with the objective of monitoring and evaluation of degradation.

Sample: A quantity of complex darunavir: β -cyclodextrin equivalent to 10 mg of darunavir was weighed accurately into a 10 mL calibrated flask; 5 mL of diluents solution was added and sonicated for 10 min to complete dissolution of the darunavir; then the mixture was diluted to the mark with the diluents to obtain a concentration of 1000 µg mL⁻¹. 1 ml of this solution was diluted in ethyl alcohol into a 25 mL calibrated flask to obtain a concentration of 40 µg mL⁻¹. A portion of the resulting mixture was withdrawn and filtered through a 0.45 µm filter to ensure the absence of particulate matter. Thus, the filtrate was injection onto the column.

Chromatographic conditions: Chromatographic analysis was carried out at ambient temperature (25°C) on a PhenomenexTM CN Luna (250 mm \times 4.6 mm, 5.0 μ m particle size) column. The mobile phase was a mixture of water + 0.1 % glacial acetic acid and acetonitrile + 0.1 % glacial acetic acid (60:40, v/v). Flow rate was 1.0 mL min⁻¹. The detector wavelength was set at 268 nm with injection volume at 20 µL. The same conditions were used to LC-MS, but without addition of acid in mobile phase. Mass spectrometry conditions: Before samples were injected into the LC-MS they were applied to preparative plate thin layer chromatography (TLC) to remove the β -cyclodextrin, using purified water and methanol 70:30 (v/v) adjusted to pH 2.4 with glacial acetic acid as mobile phase. Mass spectral analyses were performed on AmaZon SL (BrukerTM) provided with ESI ion source and ion trap mass analyzer. Analyses were carried out using nitrogen as nebulizing (60 psi) and drying gas (10 L min⁻¹, 320°C). The capillary high voltage was set to 3500 V.

Infrared conditions: In analytical balance model H51 Mettler Toledo[™] were weighed 2.5 mg of sample diluted in potassium bromide, previously dried to constant weight in an oven Nova Ética[™], to form pellets of 150 mg each. The analyzes and comparisons of the spectra were carried out in transmittance.

Conditions of thermal analysis: 1 to 3 mg of sample was weighed and placed in an alumina crucible. The temperature range used in the TG was 30 to 600°C and DSC was 25 to 150°C, both under a dynamic nitrogen atmosphere (flow rate 100 ml/min) and a heating rate of 2°C/min.

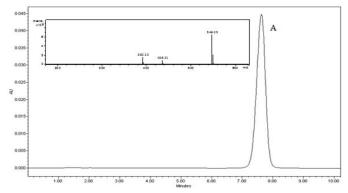
Conditions of X-ray diffraction: The X-ray diffractometer operated at 40 KV of voltage and 70 mA of tube current and. The samples were placed on a sample port of 2 cm diameter and 1 mm depth.



Results

Degradation products

The chromatograms of darunavir complexed obtained at zero time, ie before being placed in a climatic chamber showed retention time of 7.2 minutes. (Figure 1) illustrates the chromatogram of complexed drug at time zero.(Figure 2) shows the overlap of the complex chromatograms at times 3, 6, 9, 12, 18 and 24 months and (Figure 3) shows signal of the mass spectra obtained in the chromatograms in these times.





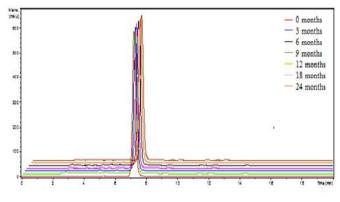
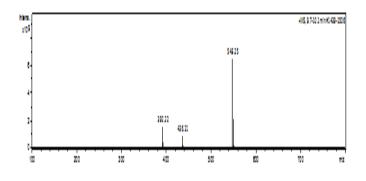


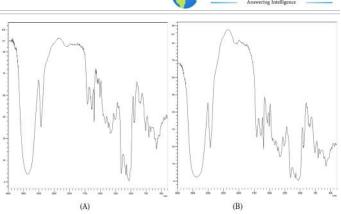
Figure 2:





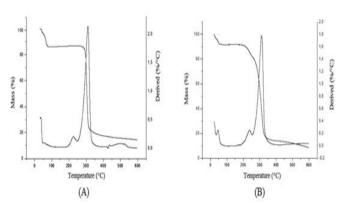
Analysis with degraded complex darunavir: β-cyclodextrin 24 months in climatic chamber

After 24 months in climate chamber the complex darunavir: β -cyclodextrin in powder was analyzed by techniques of infrared, TG, DSC and X-ray diffraction and its results were compared with zero time analyzes (no degradation), according to (Figures 4 to7).

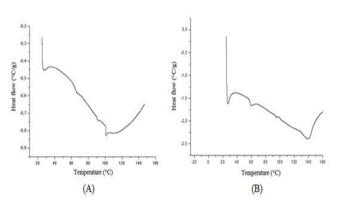


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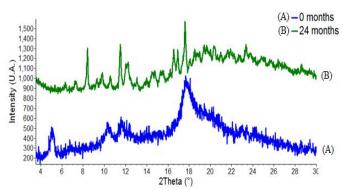






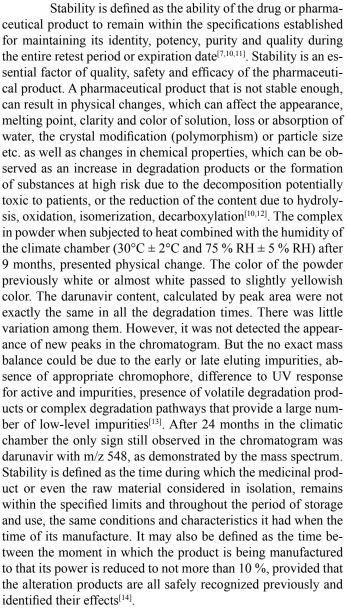








Discussion



Thus, the IC under conditions of $30 \pm 2^{\circ}$ C and 75 % RH \pm 5 % RH after 24 months is stable in powder form. For future pharmaceutical form new stability tests for the finished product would be needed. The resources of pharmaceutical technology will be required to protect the powder of the physical change, for the finished product of darunavir complexed not present color difference within 2 years or more. Another proof of the stability of the complex after the climatic chamber conditions, are the results of infrared, TG, DSC and X-ray diffraction performed on the sample submitted after 24 months at $30 \pm 2^{\circ}$ C and 75 % RH \pm 5 % UR. The results are similar to those obtained at time zero analysis, without any degradation. In infrared analysis, the inclusion complex presents a broad band between 3000 and 3750 cm-1 as β -CD, but it keeps all the characteristics bands of darunavir (primary amine, alcohol, alkanes, benzene ring, ester, amide and sulphoxide secondary), proving that the drug is present in the complex^[9,15]. The TG curve exhibits two mass loss events, the first at 50°C and the second at 300°C. In DSC still no thermal event can be observed, which should be due to changes in the crystalline solid form or by complexation with β -cyclodextrin^[9]. In the analysis of X-ray diffraction the darunavir: β -cyclodextrin complex zero time is semi-crystalline, in which we can observe some broad peaks indicating a small average size of crystalline and the characteristic halo of amorphous. In the darunavir: β -cyclodextrin complex 24 months showed an increase in crystallinity with the definition of some peaks of X-ray diffraction but also with semi-crystalline aspect. The peaks could not be attributed to the drug or cyclodextrin or indicating that darunavir: β -cyclodextrin complex is more crystalline in 24 months.

Conclusions

The study of the long-term stability of the complex darunavir: β -cyclodextrin in powder was possible through the techniques of HPLC and LC-MS. Infrared, TG, DSC and X-ray diffraction analysis confirmed the stability of the complex after 24 months at $30 \pm 2^{\circ}$ C and 75 % RH \pm 5 % UR in climatic chamber.

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Conflict of interest

The authors report no conflict of interest.

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